ENGG*3590 WATER QUALITY

School of Engineering, University of Guelph Fall 2008

Instructor:	Eyad A. Barakat, MSc, PhD Candidate (ebarakat@uoguelph.ca)	
GTA:	Lab Section 1 & 2: Fraser Kent (fkent@uoguelph.ca) Tutorial: Christopher Potvin (cpotvin@uoguelph.ca) (GTAs have no office hours; sufficient contact time in lab and tutorial)	
Lecture Times:	Tuesday and Thursday from 10:00 to 11:20 in AXEL 117	
Lab:	Monday and Tuesday from 14:30 - 17:20 in Engr. 1193. Schedule will be posted on WebCT.	
Tutorial:	Monday from 14:30 to 16:20 in Thornbrough 1103.	
Office Hours:	Tuesday @ 12:00 to 14:00 Room 207, Thornbrough. or by appointment.	
Texts/Notes:	Tchobanoglous, G. and Schroeder, E.D. 1987, <i>Water Quality</i> , Addison Wesley Longman, 768p. Zytner, R.G. 1998, <i>Water Quality Laboratory Manual</i> , SOE. PDF files on Course WebCT.	
Exams:Midterm: Final:	Thursday October 16, 2008 in AXEL 117 during the lecture slot Wednesday December 10, 2008 from 19:00 to 21:00 – Room TBA	
Prerequisites:	As stated in the U of G Calendar	
Announcements:	See course WebCT for posted materials and updates.	

COURSE SUMMARY

Water Quality is an essential course for undergraduate students in the Water Resources and Environmental Engineering programs. The concepts and principles presented give students the necessary engineering skills to address the water quality problems they will face in their senior year and upon graduation.

This course builds on the student's experience in chemistry, fluid mechanics, engineering science and provides an engineering perspective on:

- physical, chemical and biological characteristics of water
- standard methods of water quality analysis
- significance and interpretation of analytical results
- modelling of water quality in natural systems
- introduction to engineered water and wastewater treatment systems

EVALUATION

:	Quizzes Two Individual Laboratory Reports	10% 15%
:	Midterm Final Exam	35% 40%

Note: Students must attain a combined total of 50% on the examinations (Midterm and Final Exam) to pass the course. If not, that grade will be assigned for the course.

COURSE OUTLINE

I - Introduction, Sources of Water and Wastewater, Water Recycle			
 II - Physical, Chemical and Biological Characteristics of Water turbidity solids - sludge volume taste, odour and temperature pH, acidity, alkalinity and hardness ThOD, COD and BOD nutrients - eutrophication synthetic organics gases microorganisms - bacteria, viruses, pathogens, coliform, Cryptosporidium 	5 weeks		
 III - Analysis and Sampling Methods (<i>water and solid matrices</i>) physical, chemical and biological grab, composite, continuous and remote preservation gravimetric (solids), volumetric (titration), photometric (colour, nitrates, iron), electrometric temperature), culturing (coliform, plate counts) overview of high tech (GC, GC-MS, ICP, HPLC) detection limits 	0.5 week (pH, DO,		
IV - Simple river model (oxygen sag) 1 week			
 IV - Water Treatment history pretreatment - source, screens, pre-chlorination, sedimentation, aeration treatment - coagulation & sedimentation (Type I and Type II settling), filtration, ozonation, post ch overview of special treatment - activated carbon, fluoride, softening 	2.5 weeks		
 V - Wastewater Treatment history pretreatment - source, screens, bar racks treatment - sedimentation (Type III and Type IV settling), attached growth, suspended growth, wetlands septic systems overview of special treatment - tertiary treatment, BNR, membrane 			
LABORATORY EXPERIMENTS			

Laboratory work will consist of the following five water quality tests:

- biochemical oxygen demand (BOD)
- ► coagulation and flocculation (C&F)
- coliforms (total and fecal)
- chlorine demand
- ► solids fractions, Type I and/or Type II settling

Assignments:

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Approximately ten assignments will be issued throughout the term. Assistance will be available during the tutorial period (Monday from 14:30 to 16:20) to assist in solving the problems and to provide the solutions on request. **Please note that the solutions will not be posted.**

Quizzes:

There will be 4-6 short pre announced in class quizzes during the term, equally weighted worth 10% in total. The one quiz with the lowest mark will be discarded from the term evaluation. These will consist of multiple choice questions, short answers and/or calculation format. The purpose is to test you on the material as we move through the course. Quizzes schedule will be posted on WebCT. If you are not in class you will receive a zero unless there are extenuating circumstances. There will be NO makeup quizzes. Additional details will be provided in class.

Laboratory and Laboratory Reports:

Six lab sessions have been scheduled, with students working in pairs (your choice). Specific schedules will be posted by the instructor on the WebCT.. The procedures for each laboratory are outlined in the *Lab Manual*, including safety issues. Please read the appropriate sections prior to the lab, to ensure that the lab flows smoothly. If you own a lab coat, please bring it to the laboratory.

Each student will prepare two individually written lab reports for one of the lab sessions using the appropriate data set. The topic and due date will be determined randomly by the instructor. Further detail on the lab reports is given in the lab manual. Late laboratory reports will not be accepted. **There will be no exceptions.** See statement below on Academic Misconduct.

Mid-Term Exam:

The material covered will include the last Thursday lecture prior to the exam. The exam will be closed book - multiple choice/short answers and calculation format. Failure to write the exam will lead to a zero. The only exception will be for students with a medical reason signed by a physician. **There will be no exceptions.**

Final Exam:

The final exam will be comprehensive of all the material covered. Questions will be of the calculation format. Failure to attend the exam will lead to a zero for that exam. The only exception will be for students with a medical reason signed by a physician. **There will be no exceptions.**

Additional Resources:

The following books are available for additional reading and reference material:

Davis, Mackenzie L. and Cornwell, David A. 2008, Introduction to Environmental Engineering, McGraw Hill

MWH. 2005, Water Treatment Principles and Design, John Wiley & Sons, Inc.

Peavy, Howard S., Rowe, Donald R. and Tchobanglous, George. 1985, *Environmental Engineering*, McGraw Hill Publishing Company

Ramaswami, Anu, Milford, Jana B. and Small, Mitchell J. 2005, *Integrated Environmental Modeling: Pollutant Transport, Fate, and Risk in the Environment*, John Wiley & Sons, Inc.

Sawyer, Clair N., McCarty, Perry L. and Parkin, Gene F. 2003, 5th Edition, *Chemistry for Environmental Engineering* and Science, John Wiley & Sons, Inc.

PLEASE NOTE:

- The Regulations concerning Academic Misconduct as outlined in the University of Guelph, Undergraduate Calendar for 2008-2009 will be strictly enforced.
 - accordingly, when you submit your Literature Review and Laboratory Report, please include a statement that the submitted work was a solo effort. This also requires that you to provide your SMP number if you are an engineering student. Failure to include this statement and a valid SMP number will mean that your submission will not be graded.
- There will be no supplemental work for improved grades.
- The GTAs have no office hours as there is sufficient contact time in the tutorials.

DISCLAIMER

The instructor reserves the right to change any or all of the above in the event of appropriate circumstances, subject to University of Guelph Academic Regulations.

August 27, 2008