

# **ENGG\*2120** Material Science

Winter 2019 Section(s): C01

School of Engineering Credit Weight: 0.50 Version 1.00 - January 05, 2019

### 1 Course Details

# 1.1 Calendar Description

Study of the mechanical, electrical, magnetic, optical and thermal properties of solids. Atomic order and disorder in solids, single-phase metals, and multiphase materials (their equilibria and micro-structure) are examined as a basis for understanding the causes of material properties. Interwoven throughout the course is an introduction to materials selection and design considerations.

**Pre-Requisite(s):** CHEM\*1040, PHYS\*1130

### 1.2 Timetable

#### Lectures:

Tuesday & 11:30 AM – 12:50 PM ROZH Room

Thursday 103

### Laboratory: Materials Science Lab (THRN 1008)

Sec 1	Monday	9:30 AM - 11:20 AM	THRN 1008
Sec 2	Friday	8:30 AM - 10:20 AM	THRN 1008
Sec 3	Monday	2:30 PM - 4:20 PM	THRN 1008
Sec 4	Tuesday	2:30 PM - 4:20 PM	THRN 1008
Sec 5	Wednesday	y2:30 PM - 4:20 PM	THRN 1008

Sec 6 Thursday 2:30 PM - 4:20 PM THRN 1008

Sec 7 Friday 2:30 PM - 4:20 PM THRN 1008

### 1.3 Final Exam

Wednesday April 17th 2019 2:30 PM - 4:30 PM

Room to be announced.

# **2 Instructional Support**

# 2.1 Instructional Support Team

Instructor: Abdallah Elsayed Ph.D, EIT aelsay01@uoguelph.ca
Telephone: +1-519-824-4120 x56933

Office: RICH 2523

Office Hours: Tuesday and Thursday 2:00 PM – 3:00 PM, via email or by

appointment

**Lab Technician:** Barry Verspagen

Email: baverspa@uoguelph.ca
Telephone: +1-519-824-4120 x58821

Office: THRN 1138

# 2.2 Teaching Assistant(s)

**Teaching Assistant:** Fatima Haque

Email: fhaque@uoguelph.ca
Office Hours: By appointment

Teaching Assistant: Rutvik Soni

Email: sonir@uoguelph.ca
Office Hours: By appointment

**Teaching Assistant:** Parth Jivani

**Email:** pjivani@uoguelph.ca **Office Hours:** By appointment

**Teaching Assistant:** Bharathwaj Ananthapillai banantha@uoguelph.ca

Office Hours: By appointment

# **3 Learning Resources**

### 3.1 Required Resource(s)

D.R. Askeland, and W.J. Wright, The Science and Engineering of Materials, 7th Edition, SI, Cengage Learning, 2015. (Textbook)

### **Courselink (Website)**

https://courselink.uoguelph.ca

Course material, news, announcements, and grades will be regularly posted to the ENGG\*2120 Courselink site. **You are responsible for checking the site regularly.** 

## 3.2 Recommended Resource(s)

### Further reading (Textbook)

W.D. Callister and D.G. Rethwisch, Materials Science and Engineering: An Introduction, 9th Edition, John Wiley & Sons, Inc., 2014.

### Iclicker (Equipment)

Iclickers will be used to explain concepts and gauge student learning. It is recommended that you bring iclickers to class to participate in class examples. Iclicker usage will not count towards any grades.

### 3.3 Additional Resources

**Lecture Information**: Lecture notes will be posted on Courselink.

Please note that power point presentations are not comprehensive of all materials covered. During lecture, additional notes and examples will be provided.

**Lab Information**: The lab manual and schedule for the laboratory exercises are posted on Courselink.

Be sure to read the appropriate lab instructions prior to attending the lab.

**Assignments**: Study assignments will be posted at the end of a chapter or a group of chapters, with the solutions to follow about one week later. Assignments will not be marked. It is strongly recommended that you work through these assignments as they are valuable study aids and similar to the types of questions that may be asked on an exam.

# **4 Learning Outcomes**

This course is an introductory course in materials science. The student will be introduced to the atomic or molecular structure of metals, polymers, ceramics, and composite materials and learn how these different structures influence their mechanical, electrical and thermal

behaviour. Many of the differences between properties of classes of materials are related to the atomic structure of the material.

The mechanical properties of a material are influenced by the atomic arrangement and presence of crystallographic defects. In addition, methods of controlling the atomic arrangement of a material such as heat treating and strain hardening will be investigated. Finally, common service failures due to creep, fatigue, or fast fracture will be examined in light of the atomic structure of the different materials.

The course will also examine material consideration for design. Each material has its own unique properties and characteristics. Understanding how the material properties can change with the environment and how the properties can be manipulated will provide more informed material selection choices. A properly selected material can enhance a design through structural changes and greater performance while an improperly selected material can lead to complete design failure.

## **4.1 Course Learning Outcomes**

By the end of this course, you should be able to:

- 1. Describe the general properties of key engineering materials: metals, semiconductors, ceramics, polymers, and composites through a material identification project.
- 2. Create simple lab experiments to measure material properties and evaluate the effectiveness of the experiment in measuring those properties through a material identification project.
- 3. Recognize the concepts of stress, strain and how they are related, and determine the elastic modulus, the yield strength, the tensile strength and the ductility for a given engineering stress-strain curve
- 4. Recognize the link between the atomic structure of a material and its macroscopic properties through testing of material properties such as strength, stiffness, and impact behaviour.
- 5. Explain how the microstructure of a material can be manipulated by altering the operating environment, strain hardening, and heat treatment through lab report discussion questions.
- 6. Draw unit cells and derive the relationships between unit cell edge length and atomic radius for FCC, BCC and HCP crystal structures. And determine the directional and planar (Miller) indices within a unit cell.
- 7. Determine the phases present, the compositions of the phases and the mass fractions of the phases and for some given phase diagrams.

8. Present, analyze, and discuss experimental data through well written lab reports.

# 4.2 Engineers Canada - Graduate Attributes (2018)

Successfully completing this course will contribute to the following:

#	Outcome	Learning Outcome(s)
1	Knowledge Base	1, 3, 4, 5, 6, 7
1.2	Recall, describe and apply fundamental principles and concepts in natural science	1, 3, 4, 5, 6, 7
1.3	Recall, describe and apply fundamental engineering principles and concepts	1, 3, 4, 5, 6, 7
2	Problem Analysis	1
2.2	Identify, organize and justify appropriate information, including assumptions	1
3	Investigation	2, 8
3.1	Propose a working hypothesis	2, 8
3.2	Design and apply an experimental plan/investigative approach (for example, to characterize, test or troubleshoot a system)	2, 8
3.3	Analyze and interpret experimental data	2, 8
3.4	Assess validity of conclusions within limitations of data and methodologies	2, 8
5	Use of Engineering Tools	2
5.1	Select appropriate engineering tools from various alternatives	2
5.2	Demonstrate proficiency in the application of selected engineering tools	2
5.3	Recognize limitations of selected engineering tools	2
7	Communication Skills	2, 8
7.1	Identify key message(s) and intended audience in verbal or written communication as both sender and receiver	2, 8
7.2	Interpret technical documentation such as device specification sheets, drawings, diagrams, flowcharts, and pseudocode	2, 8
7.3	Construct the finished elements using accepted norms in English, graphical standards, and engineering conventions, as appropriate for the	2, 8

#		Learning Outcome(s)
	message and audience	

# **5 Teaching and Learning Activities**

Please note that the order of topics is tentative and may be adjusted at the discretion of the instructor.

### 5.1 Lecture

Lecture 1

**Topic(s):** Introduction

**Reference(s):** Chapter 1

Lecture 2, 3

**Topic(s):** Atomic Structure

Reference(s): Chapter 2

Lecture 3, 4

**Topic(s):** Mechanical Properties

**Reference(s):** Chapter 6

Lecture 5, 6

**Topic(s):** Failure Mechanisms

**Reference(s):** Chapter 7

Lecture 7, 8

**Topic(s):** Atomic Arrangement

**Reference(s):** Chapter 3

Lecture 9, 10

**Topic(s):** Imperfections in Atomic Arrangement

Reference(s): Chapter 4

Lecture 11

**Topic(s):** Ferrous & Non-Ferrous Alloys

Reference(s): Chapter 13, 14

Lecture 12

Topic(s): Ceramics

Reference(s): Chapter 15

**Lecture 13, 14** 

Topic(s): Polymers

Reference(s): Chapter 16

Lecture 15

**Topic(s):** Composites

**Reference(s):** Chapter 17

**Lecture 16, 17** 

**Topic(s):** Solid Solutions

Reference(s): Chapter 10

**Lecture 18, 19** 

**Topic(s):** Dispersion Strengthening - Phase Diagrams

Reference(s): Chapter 11

**Lecture 20, 21** 

**Topic(s):** Dispersion Strengthening - Phase Transformations

Reference(s): Chapter 12

Lecture 22

**Topic(s):** Heat Treating of Steel

Reference(s): Chapter 13

Lecture 23

**Topic(s):** Strain Hardening

Reference(s): Chapter 8

Lecture 24

**Topic(s):** Semiconductors

**Reference(s):** Chapter 19

#### 5.2 Lab Schedule

A detailed lab schedule is posted on Courselink. The labratory schedule provides information on groups, experiments and project. All lab reports must be submitted electronically in the dropbox on Courselink for marking by 4:00 PM two weeks after the laboratory is performed (unless indicated otherwise). For the weeks students are not in the lab, they are expected to be writing their lab report, or preparing for their next lab exercise. GTAs will be available during the lab time to answer questions.

### You must attend the lab section you are registered in.

A grace period of the first two weeks will be in effect for students attending the wrong lab sections. Afterwords, a penalty of 20% will be applied to the lab reports for students who do not attend the lab section for which they are registered.

## **5.3 Other Important Dates**

- Monday, January 7, 2019: First day of class
- Tuesday, January 8, 2019: First day of ENGG2120
- Monday, Februrary 18-Friday, Februrary 22, 2019: Winter break
- Friday, March 8, 2019: 40th class day, last day to drop courses
- Friday, April 5, 2019: Last day of class

See Schedule of Dates for other important dates in the academic year.

https://www.uoguelph.ca/registrar/calendars/undergraduate/2018-2019/c03/c03-wintersem.shtml

# **6 Assessments**

Students must achieve at least 50% of the marks assigned to the midterm and final exams in order for the labs to be counted in the final grade. If you do not achieve at least 50% of the marks assigned to the midterm and final exams, the weighting of the lab reports in your final grade will be zero. An overall final grade of 50% is required to pass the course.

# 6.1 Marking Schemes & Distributions

ltem	Scheme A(%)	Scheme B (%)
Lab Reports	20	20
Project	7.5	7.5
Midterm	27.5	32.5
Exam	45	40

### 6.2 Assessment Details

### Assignments (0%)

Learning Outcome(s): 3,6,7

Study assignments will be posted at the end of a chapter or a group of chapters, with the solutions to follow about one week later. Assignments will not be marked. It is strongly recommended that you work through these assignments as they are valuable study aids and similar to the types of questions that may be asked on an exam.

#### Lab Reports (20%)

Learning Outcome(s): 4,5,8

See the Lab Schedule for schedule and report due dates.

### **Materials Identification Project (7.5%)**

**Learning Outcome(s):** 1,2

See the Lab Schedule for Project schedule and report due dates.

Projects are to be completed with the same members as your lab groups.

### Midterm (27.5%)

**Learning Outcome(s):** 3,6,7 Februrary 28, 2019 in class

Each student is allowed one **single-sided** 8.5" x 11" note sheet for the exam. Each note sheet must be prepared by you (typed or handwritten) and be your own original work (i.e. not a copy). Numerical solutions or steps for solving problems are not allowed.

#### Final Exam (45%)

Date: Wed, Apr 17, 2:30 PM - 4:30 PM, Room TBD

Learning Outcome(s): 3,6,7

Final Exam location is set by the University Registrar.

Each student is allowed one **double-sided** 8.5" x 11" note sheet for the exam. Each note sheet must be prepared by you (typed or handwritten) and be your own original work (i.e. not a copy). Numerical solutions or steps for solving problems are not allowed.

## 7 Course Statements

### 7.1 Introduction

- 1. Sharing of calculators, formula sheets, if applicable, or use of smart phones as calculators is not allowed.
- 2. Grading is based on the procedure, correctness of numerical calculations and final answer.
- 3. The instructor, at his discretion, may entertain requests by the class to adjust assessment dates, except final exam, with the unanimous consent of the class.

### 7.2 Grading Policies

**Missed Assessments**: If you are unable to meet an in-course requirement due to medical, psychological, or compassionate reasons, please email the course instructor. See the undergraduate calendar for information on regulations and procedures for Academic Consideration: http://www.uoguelph.ca/registrar/calendars/undergraduate/current/c08/c08-ac.shtml

**Accommodation of Religious Obligations**: If you are unable to meet an in-course requirement due to religious obligations, please email the course instructor within the first two weeks of the first class to make alternate arrangements. See the undergraduate calendar for information on regulations and procedures for Academic Accommodation of Religious Obligations: http://www.uoguelph.ca/registrar/calendars/undergraduate/current/c08/c08-accomrelig.shtml

**Passing Grade**: Students must achieve at least 50% of the marks assigned to the midterm and final exams in order for the labs and quizzes to be counted in the final grade. If you do not achieve at least 50% of the marks assigned to the midterm and final exams, the weighting of the lab reports and project in your final grade will be zero. An overall final grade of 50% is required to pass the course.

**Missed Midterm Exam**: If you miss a test due to grounds for granting academic consideration or religious accommodation, the weight of the missed test will be added to the final exam. There will be no makeup midterm exam.

**Remarking of Midterm Exam:** Consideration for remarking of the midterm exam will only be allowed if brought to the attention of the instructor within two weeks of when midterm

results are released.

Lab Work: You must attend and complete all labs. Doors to the lab will be closed 15 minutes after the scheduled lab time. Students arriving after the lab doors are closed are considered absent. If you miss a lab due to grounds for granting academic consideration or religious accommodation, arrangements must be made with the teaching assistant to complete a makeup lab. Unless academic consideration is granted, failure to complete a lab will result in a mark of zero for that lab report. Check your lab section. You are only allowed to attend the section you are registered in.

The laboratory work is group based. You will need to organize yourselves into groups of three (3) or four (4) within your lab section by Monday, January 14th. Be sure to choose your lab partners wisely! The sign-up sheets for lab groups will be available in the Materials Lab in THRN 1008 during the introductory lab session. You will not be allowed to conduct the project or labs unless you attend the safety session and sign a form indicating that you have done so.

Each group will be responsible for conducting the labs and writing a single report for each lab. You will be equally responsible for your group's laboratory reports. Each group member must make a significant contribution to the writing of the lab report and sign the lab report cover page in order to receive a lab report mark. Lab reports will be marked and the marks posted on Courselink. Note that up to 20% of the lab mark may be deducted for poor lab report format, poor graph or table format, or poor English (spelling, grammar, etc.). Any reports judged to be entirely unacceptable will be returned without marking for rewriting. If you have questions about your mark, see the GTA responsible for that lab and they will discuss it with you.

**Late Lab Reports**: There will be a late penalty of 20%/day or part thereof for any late lab reports. That is, reports submitted within 24 hours after the initial due date will lose 20%, reports submitted between 24 and 48 hours after the initial due date will lose 40%, and so on. Lab reports are considered late if they are submitted after the specified time they are due.

# **8 School of Engineering Statements**

# 8.1 Instructor's Role and Responsibility to Students

The instructor's role is to develop and deliver course material in ways that facilitate learning for a variety of students. Selected lecture notes will be made available to students on Courselink but these are not intended to be stand-alone course notes. Some written lecture notes will be presented only in class. During lectures, the instructor will expand and explain the content of notes and provide example problems that supplement posted notes.

Scheduled classes will be the principal venue to provide information and feedback for tests and labs.

# 8.2 Students' Learning Responsibilities

Students are expected to take advantage of the learning opportunities provided during lectures and lab sessions. Students, especially those having difficulty with the course content, should also make use of other resources recommended by the instructor. Students who do (or may) fall behind due to illness, work, or extra-curricular activities are advised to keep the instructor informed. This will allow the instructor to recommend extra resources in a timely manner and/or provide consideration if appropriate.

### 8.3 Lab Safety

Safety is critically important to the School and is the responsibility of all members of the School: faculty, staff and students. As a student in a lab course you are responsible for taking all reasonable safety precautions and following the lab safety rules specific to the lab you are working in. In addition, you are responsible for reporting all safety issues to the laboratory supervisor, GTA or faculty responsible.

# 9 University Statements

### 9.1 Email Communication

As per university regulations, all students are required to check their e-mail account regularly: e-mail is the official route of communication between the University and its students.

# 9.2 When You Cannot Meet a Course Requirement

When you find yourself unable to meet an in-course requirement because of illness or compassionate reasons please advise the course instructor (or designated person, such as a teaching assistant) in writing, with your name, id#, and e-mail contact. The grounds for Academic Consideration are detailed in the Undergraduate and Graduate Calendars.

Undergraduate Calendar - Academic Consideration and Appeals https://www.uoguelph.ca/registrar/calendars/undergraduate/current/c08/c08-ac.shtml

Graduate Calendar - Grounds for Academic Consideration https://www.uoguelph.ca/registrar/calendars/graduate/current/genreg/index.shtml

# 9.3 Drop Date

Courses that are one semester long must be dropped by the end of the fortieth class day; two-semester courses must be dropped by the last day of the add period in the second semester. The regulations and procedures for course registration are available in the Undergraduate and Graduate Calendars.

Undergraduate Calendar - Dropping Courses https://www.uoguelph.ca/registrar/calendars/undergraduate/current/c08/c08-drop.shtml

Graduate Calendar - Registration Changes https://www.uoguelph.ca/registrar/calendars/graduate/current/genreg/genreg-reg-regchq.shtml

### 9.4 Copies of Out-of-class Assignments

Keep paper and/or other reliable back-up copies of all out-of-class assignments: you may be asked to resubmit work at any time.

## 9.5 Accessibility

The University promotes the full participation of students who experience disabilities in their academic programs. To that end, the provision of academic accommodation is a shared responsibility between the University and the student.

When accommodations are needed, the student is required to first register with Student Accessibility Services (SAS). Documentation to substantiate the existence of a disability is required; however, interim accommodations may be possible while that process is underway.

Accommodations are available for both permanent and temporary disabilities. It should be noted that common illnesses such as a cold or the flu do not constitute a disability.

Use of the SAS Exam Centre requires students to book their exams at least 7 days in advance and not later than the 40th Class Day.

More information can be found on the SAS website https://www.uoguelph.ca/sas

# 9.6 Academic Integrity

The University of Guelph is committed to upholding the highest standards of academic integrity, and it is the responsibility of all members of the University community-faculty, staff, and students-to be aware of what constitutes academic misconduct and to do as much as possible to prevent academic offences from occurring. University of Guelph students have the responsibility of abiding by the University's policy on academic misconduct regardless of their location of study; faculty, staff, and students have the responsibility of supporting an environment that encourages academic integrity. Students need to remain aware that instructors have access to and the right to use electronic and other means of detection.

Please note: Whether or not a student intended to commit academic misconduct is not relevant for a finding of guilt. Hurried or careless submission of assignments does not excuse students from responsibility for verifying the academic integrity of their work before submitting it. Students who are in any doubt as to whether an action on their part could be construed as an academic offence should consult with a faculty member or faculty advisor.

Undergraduate Calendar - Academic Misconduct https://www.uoguelph.ca/registrar/calendars/undergraduate/current/c08/c08-amisconduct.shtml

Graduate Calendar - Academic Misconduct https://www.uoguelph.ca/registrar/calendars/graduate/current/genreg/index.shtml

# 9.7 Recording of Materials

Presentations that are made in relation to course work - including lectures - cannot be recorded or copied without the permission of the presenter, whether the instructor, a student, or guest lecturer. Material recorded with permission is restricted to use for that course unless further permission is granted.

### 9.8 Resources

The Academic Calendars are the source of information about the University of Guelph's procedures, policies, and regulations that apply to undergraduate, graduate, and diploma programs.

Academic Calendars https://www.uoguelph.ca/academics/calendars